INORGANIC COMPOUNDS

Acta Cryst. (1994). C50, 160-161

Refinement of Hexagonal BaTiO₃

J. AKIMOTO, Y. GOTOH AND Y. OOSAWA

National Institute of Materials and Chemical Research, Higashi, Tsukuba, Ibaraki 305, Japan

(Received 17 May 1993; accepted 13 August 1993)

Abstract

Pure single crystals of barium titanate have been synthesized and the structure refined using the atomic coordinates of Burbank & Evans [Acta Cryst. (1948), 1, 330-336]. The short Ti—Ti distance in the face-sharing Ti₂O₉ group was determined to be 2.690 (4) Å, which is significantly shorter than in other barium titanate compounds.

Comment

The crystal structure of hexagonal BaTiO₃ was originally determined by Burbank & Evans (1948), using crystals synthesized by Matthias (1948) that had a light amber colour, attributed to dissolved platinum from the reaction vessel (Dickson, Katz & Ward, 1961; Tillmanns, Hofmeister & Baur, 1985). The hexagonal BaTiO₃ structure can be obtained by the addition of small amounts of other octahedral cations (Dickson, Katz & Ward, 1961). The substitution of Ti by transition elements in the double group of face-sharing octahedra has been shown to give significantly short metal-metal distances, *e.g.* 2.655 (1) Å in Ba(Ti_{0.88}Pt_{0.12})O₃ (Fischer & Tillmanns, 1981). Pure single crystals of hexagonal BaTiO₃ have been obtained

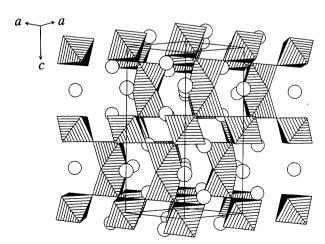


Fig. 1. STRUPLO90 (Fischer, le Lirzin, Kassner & Rüdinger, 1991) drawing of BaTiO₃ viewed along [110].

©1994 International Union of Crystallography Printed in Great Britain – all rights reserved recently by partial melt growth at 1853 K, below the melting temperature, using a molybdenum crucible (Akimoto, Gotoh, Sohma, Kawaguchi & Oosawa, 1994). The chemical composition was determined using an energy-dispersive X-ray analyser. No contamination, such as by molybdenum from the crucible, was observed. The cell parameters are consistent with the powder data (JCPDS File No. 34-129). The length of the *c* axis is slightly shorter than those of 14.05 (Burbank & Evans, 1948) and 14.023 (1) Å (Fischer & Tillmanns, 1981) found in Ba(Ti,Pt)O₃ compounds.

Experimental

Crystal data BaTiO₃ $M_r = 233.2$ Hexagonal $P6_3/mmc$ a = 5.7238 (7) Å c = 13.9649 (7) Å V = 396.22 (9) Å³ Z = 6 $D_x = 5.865$ Mg m⁻³

Data collection

Rigaku AFC-5S diffractometer

 $\omega/2\theta$ scans [ω -scan width (1.0 + 0.5tan θ)° and rate 2.0° min⁻¹]

Absorption correction: by integration from crystal shape (ACACA; Wuensch & Prewitt, 1965) $T_{min} = 0.257, T_{max} =$ 0.428

Refinement

Refinement on F R = 0.037 wR = 0.036 591 reflections 22 parameters Weighting scheme based on measured e.s.d.'s, w = $1/\sigma^2(F)$ $(\Delta/\sigma)_{max} = <0.01$

Mo $K\alpha$ radiation $\lambda = 0.71069$ Å Cell parameters from 25 reflections $\theta = 25-35^{\circ}$ $\mu = 17.57 \text{ mm}^{-1}$ T = 296 KPlate $0.100 \times 0.088 \times 0.051 \text{ mm}$ Colourless

1280 measured reflections 676 independent reflections 591 observed reflections $[F > 3\sigma(F)]$ $\theta_{max} = 45^{\circ}$ $h = 0 \rightarrow 5$ $k = 0 \rightarrow 9$ $l = 0 \rightarrow 27$ 3 standard reflections monitored every 50 reflections intensity variation: 0.98%

```
\Delta \rho_{\text{max}} = 3.7 \text{ e } \text{\AA}^{-3}\Delta \rho_{\text{min}} = -6.8 \text{ e } \text{\AA}^{-3}Extinction correction:
Coppens & Hamilton
(1970)
Extinction coefficient:
g = 0.96 (3) \times 10^{-4}Atomic scattering factors
from Cromer & Mann
(1968)
```

Data collection: Rigaku AFC software. Cell refinement: Rigaku AFC software. Program(s) used to refine structure: *RFINE2* (Finger, 1969). Molecular graphics: *STRUPLO*90 (Fischer, le Lirzin, Kassner & Rüdinger, 1991).

Acta Cryst. (1994). C50, 161-164

Single-Crystal X-ray Diffraction

Center for Superconductivity Research and

Department of Chemistry and Biochemistry. University of Maryland, Maryland 20742, USA

National Institute of Standards and Technology,

(Received 26 March 1993; accepted 24 August 1993)

BAI-HAO CHEN AND BRYAN EICHHORN

Gaithersburg, MD 20899, USA

WINNIE WONG-NG

Structural Reinvestigation of Ba₃Zr₂S₇ by

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ($Å^2$)

$\mathbf{D}_{\mathrm{eq}} = (\mathbf{q}, \mathbf{y}) \mathbf{D}_{\mathrm{f}} \mathbf{D}_{\mathrm{f}} \mathbf{a}_{\mathrm{f}} \mathbf{a}_{\mathrm{f}}.$					
x	у	z	Beq		
0	0	1/4	0.43		
1/3	2/3	0.09671 (5)	0.52		
0	0	0	0.67		
1/3	2/3	0.84633 (14)	0.62		
0.5185 (6)	0.0370	1/4	0.65		
0.8349 (6)	0.6698	0.0802 (2)	0.66		
	1/3 0 1/3 0.5185 (6)	x y 0 0 1/3 2/3 0 0 1/3 2/3 0.5185 (6) 0.0370	x y z 0 0 1/4 1/3 2/3 0.09671 (5) 0 0 0 1/3 2/3 0.84633 (14) 0.5185 (6) 0.0370 1/4		

Table 2. Selected geometric parameters (Å, °)

Ti(1)O ₆ octahedro	n					
O(2)—O(2)	3.965 (4)	× 3	O(2) - Ti(1) - O(2)	180.00 (1)		
O(2)—O(2)	2.8334 (3)	× 6	O(2) - Ti(1) - O(2)	91.2 (1)		
O(2)—O(2)	2.774 (7)	× 6	O(2) - Ti(1) - O(2)	88.8 (2)		
Ti(1)O(2)	1.983 (2)	× 6				
Ti(1)—Ti(2)	3.940(1)					
Ti(2)O ₆ octahedron						
O(1) - O(2)	3.932 (2)	× 3	O(1) - Ti(2) - O(2)	169.1 (1)		
O(1) - O(1)	2.5434 (3)	× 3	O(1) - Ti(2) - O(1)	79.37 (6)		
O(2)O(2)	2.8904 (3)	× 3	O(2) - Ti(2) - O(2)	95.1 (1)		
O(1)-O(2)	2.848 (4)	× 6	O(1) - Ti(2) - O(2)	92.3 (1)		
Ti(2)—O(1)	1.991 (1)	× 3				
Ti(2)—O(2)	1.959 (2)	× 3				
Ti(2)—Ti(2)	2.690 (4)					
Ba-O cuboctahedra						
Ba(1)O(1)	2.868 (4)	× 6	Ba(2) - O(1)	2.820 (6) × 3		
Ba(1)—O(2)	2.881 (3)	× 6	Ba(2)—O(2)	2.871 (4) × 6		
			Ba(2)—O(2)	2.981 (3) × 3		

Lists of structure factors, anisotropic displacement parameters and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71575 (11 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: OH1053]

References

- Akimoto, J., Gotoh, Y., Sohma, M., Kawaguchi, K. & Oosawa, Y. (1994). J. Solid State Chem. In the press.
- Burbank, R. D. & Evans, H. T. Jr (1948). Acta Cryst. 1, 330-336.
- Coppens, P. & Hamilton, W. C. (1970). Acta Cryst. A26, 71-83.
- Cromer, D. T. & Mann, J. B. (1968). Acta Cryst. A24, 321-324.
- Dickson, J. G., Katz, L. & Ward, R. (1961). J. Chem. Soc. 83, 3026-3029
- Finger, L. W. (1969). Carnegie Inst. Washington Yearb. 67, 216-217.
- Fischer, R. & Tillmanns, E. (1981). Z. Kristallogr. 157, 69-81.
- Fischer, R. X., le Lirzin, A., Kassner, D. & Rüdinger, B. (1991). Z. Kristallogr. Suppl. 3, p. 75.
- Matthias, B. (1948). Phys. Rev. 73, 808.
- Tillmanns, E., Hofmeister, W. & Baur, W. H. (1985). J. Solid State Chem. 58, 14-28.
- Wuensch, B. J. & Prewitt, C. T. (1965). Z. Kristallogr. 122, 24-59.

Abstract

Single crystals of barium zirconium sulfide, Ba₃Zr₂S₇, were prepared from BaS, Zr and S at 1323 K by using a BaCl₂ flux. The compound belongs to the Ruddlesden-Popper $(A_{n+1}B_nX_{3n+1})$ family with A = Ba, B = Zr, X = Sand n = 2. It is isostructural to Sr₃Ti₂O₇ and crystallizes in alternately stacked double perovskite BaZrS₃ layers and rocksalt BaS layers.

Comment

A number of new Ruddlesden-Popper $A_{n+1}B_nX_{3n+1}$ compounds (Ruddlesden & Popper, 1957, 1958) have been discovered recently where A = Ba, B = Zr, Hf and X = S. The previous studies have shown that the Ba_2MS_4 phases (n = 1) are body-centered tetragonal, *I4/mmm* (No. 139) (Chen & Eichhorn, 1991; Saeki, Yajima & Onoda, 1991), the Ba₃Zr₂S₇ compound (n = 2) is C-centered orthorhombic, Cccm (No. 66) (Saeki, Yajima & Onoda, 1991), the n = 3, 4, 5 members are face-centered orthorhombic, Fmmm (No. 69) (Chen, Eichhorn & Fanwick, 1992; Chen, Wong-Ng & Eichhorn, 1993), and the BaMS₃ $(n = \infty)$ compounds are orthorhombic, Pnma (No. 62) (Clearfield, 1963; Lelieveld & IJdo, 1980).

A review of these structures reveals that the lower nmembers of the Ruddlesden-Popper sulfides have high crystal symmetry (I4/mmm) whereas the higher *n* members ($\infty > n > 2$) display lower *Fmmm* crystal symmetry. The exception is the n = 2 member, Ba₃Zr₂S₇ (Saeki, Yajima & Onoda, 1991), which was shown by electron diffraction and Rietveld profile refinement to have Cccm symmetry. This phase was prepared by a different synthetic method and displayed trends in bond lengths at odds with those of the other members of the $Ba_{n+1}M_nS_{3n+1}$ series (Chen, Wong-Ng & Eichhorn, 1993). Therefore, a reinvestigation of the synthesis and structure (singlecrystal X-ray diffraction) of Ba₃Zr₂S₇ was undertaken to address this structural discrepancy.

161